

Detailed Syllabus
Lecture-wise Breakup

Course Code	19M21PH115	Semester: EVEN	Semester: 2nd Session 2019-20 Month from: Jan to June
Course Name	Classical Electrodynamics		
Credits	4	Contact Hours	3+1
Faculty (Names)	Coordinator	Anirban Pathak	
	Teacher	Anirban Pathak	
COURSE OUTCOMES			COGNITIVE LEVELS
CO1	Recall basics of electrostatics, magnetostatics and electrodynamics		Remembering (C1)
CO2	Explain various physical phenomena and working of devices which involve the basic principles of electrostatics and electrodynamics		Understanding (C2)
CO3	Apply the laws of electrostatics and Maxwell's equation to solve boundary value problems and problems related to communication.		Applying (C3)
CO4	Analyze complex physical problem of relativistic and nonrelativistic electrodynamics		Analyzing (C4)
Module No.	Title of the Module	Topics in the Module	No. of Lectures for the module
1.	Electrostatics and boundary value problems in electrostatics	Coulomb's law, Gauss's law, Laplace and Poisson equations, Method of Images, Boundary value problems (in spherical and cylindrical coordinates), multipole expansion and Dielectrics (energy and forces in dielectric systems), minimum energy theorem, applications of electrostatic fields (e.g., electrostatics particle precipitators, photo duplication or Xerography and electrostatic lenses).	9
2.	Magnetostatics, Faraday's law and quasi-static fields	Biot-Savart law, differential equation of magnetostatics and Ampere's law, Faraday's law of induction, Magnetic vector potential, multipole expansion of the vector potential and magnetic field in matter, energy in the magnetic field	6
3.	Electrodynamics	Time varying field, continuity equation, Maxwell's equations, Pointing theorem, Gauge transformations, gauge invariance, Electromagnetic waves in free space, dielectrics and conductors, Fresnel's equations,	12
4.	Radiation and special theory of relativity	Retarded potentials, Lienard-Wiechert Potentials, fields due to a Point charge moving with constant velocity, Fields due to accelerated point charge, Recollection of the ideas of special theory of relativity, Four-vector and Lorentz transformation in four dimensional space; Lorentz invariants of electromagnetic fields; Transformation of electric and magnetic field vectors.	13
Total number of Lectures			40
Evaluation Criteria			
Components	Maximum Marks		
T1	20		

T2	20
End Semester Examination	35
TA	25 [2 Quiz (10 M), Attendance (10 M) and Cass performance (5 M)]
Total	100
Recommended Reading material: Author(s), Title, Edition, Publisher, Year of Publication etc. (Text books, Reference Books, Journals, Reports, Websites etc. in the IEEE format)	
1.	J D Jackson, Classical Electrodynamics, Wiley, New Delhi 2017
2.	D. J. Griffiths, Introduction to electrodynamics, Pearson (Prentice Hall), New Delhi 2008
3.	T L Chow, Introduction to Electromagnetic Theory: A modern perspective, Jones and Bartlett Learning, New Delhi, 2014

Detailed Syllabus
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Course Code	19M21PH116	Semester: Even	Semester: II Session 2019 -2020 Month from: January to June
Course Name	Atomic, Molecular and Laser Physics		
Credits	4	Contact Hours	3+1
Faculty (Names)	Coordinator	Dr Navneet K Sharma	
	Teacher	Dr Navneet K Sharma	
COURSE OUTCOMES			COGNITIVE LEVELS
CO1	Recall basics of one, two and many electron systems. Normal & anomalous Zeeman, Paschen-Back and Stark effects; L-S and J-J coupling schemes. Hartree-Fock approximation		Remembering (C1)
CO2	Explain Born-Oppenheimer approximation. States for hydrogen molecule and molecular ion (H_2 , H_2^+). Term symbol for simple molecules		Understanding (C2)
CO3	Apply concepts of rotational spectra, vibrational spectra, electronic spectra of diatomic molecules; Franck-Condon principle. Raman spectra. Electron Spin Resonance. Nuclear Magnetic Resonance		Applying (C3)
CO4	Analyze spontaneous and stimulated emissions in laser; optical pumping population inversion, rate equations. Different laser systems like Ruby, He-Ne, CO_2 and Nd:YAG lasers		Analyzing (C4)
Module No.	Title of the Module	Topics in the Module	No. of Lectures for the module
1.	Atomic Physics	Hydrogen atom and other one electron systems, two electron systems, many electron systems, spin-orbit term, intensity of fine structure lines. Effect of magnetic and electric fields: Normal and anomalous Zeeman, Paschen-Back and Stark effects. Interaction energy in L-S and J-J coupling schemes, Lande interval rule. Hartree-Fock approximation.	10
2.	Molecular Structure	Molecular electronic states, Born-Oppenheimer approximation. States for hydrogen molecule and molecular ion (H_2 , H_2^+). Spectroscopic terms symbol for simple molecules.	8
3.	Molecular Spectra	Rotational spectra of diatomic molecules-rigid and non-rigid rotors, isotope effect, Vibrational spectra of diatomic molecules- harmonic and anharmonic vibrators, Intensity of spectral lines, dissociation energy, vibration-rotation spectra, Electronic spectra of diatomic molecules-vibrational structure of electronic transitions. Rotational structure of electronic bands (Fine structure)-P,Q, R branches, Fortrat diagram. Intensities in electronic bands-The Franck-Condon principle. Raman spectra, X-ray emission spectra. Electron Spin Resonance. Nuclear Magnetic Resonance.	14
4.	Lasers	Introduction to Laser and Maser, spontaneous and stimulated emissions, Einstein A & B coefficients, optical pumping, population inversion, rate equations, modes of resonators and coherence length, Ruby, He-Ne, CO_2 and	8

	Nd:YAG lasers.	
Total number of Lectures		40
Evaluation Criteria		
Components	Maximum Marks	
T1	20	
T2	20	
End Semester Examination	35	
TA	25 [2 Quiz (10 M), Attendance (10 M) and Cass performance (5 M)]	
Total	100	

Recommended Reading material: Author(s), Title, Edition, Publisher, Year of Publication etc. (Text books, Reference Books, Journals, Reports, Websites etc. in the IEEE format)	
1.	Physics of Atoms and Molecules by B. H. Bransden and C. J. Jochain (2 nd Ed., Pearson Education, 2003)
2.	Atomic Spectra and Atomic Structure by G. Herzberg (Dover Publications, 2003)
3.	Atoms, Molecules and Photons by W. Demtroder (Springer, 2006)
4.	Fundamentals of Molecular Spectroscopy by C. N. Banwell (McGraw Hill, 1983)
5.	Basic atomic & Molecular Spectroscopy by J. M. Hollas(Royal Society of Chemistry, 2002)
6.	Principles of Lasers by O. Svelto (5 th Ed., Springer, 2010)

Detailed Syllabus
Lecture-wise Breakup

Course Code	19M21PH117	Semester: Even	Semester: 2nd 2019-2020 Session Month from: January to June
Course Name	Statistical Mechanics		
Credits	4	Contact Hours	3+1
Faculty (Names)	Coordinator	Dr. Navendu Goswami	
	Teacher	Dr. Navendu Goswami	
COURSE OUTCOMES			COGNITIVE LEVELS
CO1	Define the basic laws and parameters related to Thermodynamics and Statistical Mechanics.		Remember Level (C1)
CO2	Explain the concepts of different Thermodynamic and Statistical Systems and Ensembles.		Understand Level (C2)
CO3	Apply the concepts of Thermodynamics and Statistical ensembles to conclude its properties.		Apply Level (C3)
CO4	Evaluating the behavior of equilibrium, non-equilibrium or a random process on the basis of suitable thermodynamic parameters, distribution functions and phase transition.		Evaluate Level (C5)
Module No.	Title of the Module	Topics in the Module	No. of Lectures for the module
1.	Evaluation of Energy States	Micro- and Macro-states, Degenerate and Non-degenerate states, Two State Systems, Harmonic Oscillators, Einstein's Model of Crystalline Solid, Density of States, Particle in a box.	6
2.	Classical Statistical Mechanics	Classical Phase space, Number of Microstates, Ideal gas, Entropy: Gibbs' Paradox, Liouville's Theorem in Classical Statistical Mechanics,	5
3.	Ensembles and Distribution Functions	Micro-canonical, canonical and grand-canonical ensembles and partition functions; Free energy and its connection with thermodynamic quantities; Classical and quantum statistics. Boltzmann Limit, Sackur-Tetrode equation.	10
4.	Applications of Distribution Functions	Degenerate Fermi gas; Ideal Bose and Fermi gases; Principle of detailed balance. Blackbody radiation and Planck's distribution law; Bose-Einstein condensation, Diamagnetism, paramagnetism, and ferromagnetism, White Dwarf Stars, Saha-Ionsization Equation.	10
5.	Phase Transition and Stochastic Processes	First- and second-order phase transitions. phase equilibria, critical point. Introduction to nonequilibrium processes, Ising model. Diffusion equation. Random walk and Brownian motion.	9
Total number of Lectures			40
Evaluation Criteria			
Components	Maximum Marks		
T1	20		
T2	20		
End Semester Examination	35		
TA	25 [2 Quiz (10 M), Attendance (10 M) and Cass performance (5 M)]		

Total	100
Recommended Reading material: Author(s), Title, Edition, Publisher, Year of Publication etc. (Text books, Reference Books, Journals, Reports, Websites etc. in the IEEE format)	
1.	Frederick Reif , <i>Fundamentals of Statistical and Thermal Physics</i> , Waveland Pr Inc, 2008.
2.	Kerson Huang , <i>Statistical Mechanics</i> , Wiley, 2 nd Ed., 1987.
3.	R K Pathria, Paul D. Beale, <i>Statistical Mechanics</i> , Academic Press, 3 rd Ed., 2011.
4.	Statistical Mechanics, Richard P. Feynman, Westview Press, USA, 2008
5.	Statistical Mechanics: An Elementary Outline (Rev.Ed.), Avijit Lahiri, Universities Press, 2015

Detailed Syllabus
Lecture-wise Breakup

Course Code	19B21PH118	Semester: EVEN	Semester: 2nd Session: 2019 -2020 Month from: Jan to June
Course Name	Condensed Matter Physics		
Credits	4	Contact Hours	3+1
Faculty (Names)	Coordinator	Prof. R.K. Dwivedi	
	Teacher	Prof. R.K. Dwivedi	
COURSE OUTCOMES			COGNITIVE LEVELS
CO1	To recall the basic concept of space lattice, lattice type and crystal structure, Bonding, band diagrams, heat capacity, thermal expansion, thermal and electronic conduction in solids like metals, semiconductors,, dielectrics, magnetics and superconductors.		Remembering (C1)
CO2	To Illustrate the Debye and Einstein's model, Croning-Penny model and various physical phenomena with interpretation based on the mathematical expressions involved.		Understanding (C2)
CO3	Apply the concepts/principles to solve the problems related to Solid State Physics.		Applying (C3)
CO4	Analyze and examine the crystal structure of solids, thermal, electrical and electronic properties and establish a correlation between structure and properties		Analyze level
Module No.	Title of the Module	Topics in the Module	No. of Lectures for the module
1.	Crystal Physics	Concept of space lattice, basis and primitive cell, bravais lattices, Crystal Symmetry, intercepts of plane and miller indices, lattice type, packing efficiency, reciprocal lattice, structure factor, crystal structures (NaCl, CsCl, Diamond and cubic ZnS) and Brag's law and X-ray diffraction methods. Bonding in solids, ionic bonding, Cohesive energy and Madelung Constant in ionic crystals.	12
2.	Thermal Properties	Phonon heat capacity, specific heat, Density of states in one dimension, Density of states in three dimension, Debye's model for density of space, Debye T ³ Law, Einstein model density of states, Thermal conductivity, A brief introduction to Imperfections.	10
3.	Band theory of solids	Free electron model, Origin of energy gap, Bloch theorem, Croning-Penney model, Wave function of electron in a periodic potential, Energy band formation in solids, Classification of solids into metals, semiconductors and insulators.	8
	Electrical properties in solids	Electrical conduction in metals and semiconductors, Intrinsic and Extrinsic semiconductors, mobility, Intrinsic carrier concentration, impurity diffusion, Carrier concentration of n-type and p-type semiconductors.	5
4.	Superconductivity	Occurrence of superconductivity, Meissner effect, Type –I and Type-II superconductor, Heat capacity, Energy gap, Isotope effect. Microwave and Infrared properties. London	5

		equations and BCS theory.	
Total number of Lectures			40
Evaluation Criteria			
Components		Maximum Marks	
T1		20	
T2		20	
End Semester Examination		35	
TA		25 [2 Quiz (10 M), Attendance (10 M) and Cass performance (5 M)]	
Total		100	
Recommended Reading material: Author(s), Title, Edition, Publisher, Year of Publication etc. (Text books, Reference Books, Journals, Reports, Websites etc. in the IEEE format)			
1.	Solid State Physics by A. J. Dekker		
2.	Solid State Physics by Charles Kittel, Wiley Publication, Eight Edition (2017)		
3.	Solid State Physics by N. W. Ashcroft & N. D. Mermin		
4.	Solid State Physics by S.O. Pillai, New Age Publications (Revised sixth Ed. (2007)		

Detailed Syllabus
Lab-wise Breakup

Course Code	19M25PH112	Semester: EVEN	Semester: 2nd Session: 2019 -2020 Month from: Jan to June
Course Name	Laboratory-2		
Credits	4	Contact Hours	8

Faculty (Names)	Coordinator(s)	B. C. Joshi
	Teacher(s) (Alphabetically)	Dinesh Tripathi and Himanshu Pandey

COURSE OUTCOMES		COGNITIVE LEVELS
CO1	Recall components of electronic circuits used in the experiments.	Remembering (C1)
CO2	Explain key applications of electronic circuits and devices used in the experiments.	Understanding (C2)
CO3	Model the circuits using electronic components and perform the experiments.	Applying (C3)
CO4	Analyze the data obtained and calculate the error.	Analyzing (C4)
CO5	Interpret and justify the results.	Evaluating (C5)

Module No.	Title of the Module	List of Experiments	CO
1.	Electronics	<ol style="list-style-type: none"> 1. To assemble a two stage common emitter RC coupled amplifier and to measure the gain as a function of frequency and hence find the gain band width 2. Design and realize Inverting and Non-inverting amplifier using 741 Op-amp. 3. To design and test the performance of an integrator using 741 OP AMP 4. To study and calculate the frequency of oscillations of Colpitts oscillator. 5. To study and calculate frequency of oscillations of OP-AMP based Hartley Oscillator. 6. Design of an RC Phase Shift Oscillator (Using IC 741 OP AMP) and calculation of its frequency of oscillation. 7. To design and set up (a) half adder & half subtractor and (b) full adder & full subtractor using NAND gate. 8. To study the single stage amplifiers; using BJT in common emitter (CE) configuration and to learn its application as a 	1-5

		<p>small signal amplification.</p> <p>9. To use the operational amplifier as filters of different frequency range.</p> <p>10. Design and study of regulated power supply.</p> <p>11. FET and MOSFET characteristics and its applications as amplifier.</p> <p>Besides above experiments, students will be trained in mechanical workshop. (Training on lathe and grinding, drilling and threading etc.)</p>	
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Evaluation Criteria	
Components	Maximum Marks
Mid Term Viva (V1)	20
End Term Viva (V2)	20
D2D	60
Total	100
Recommended Reading material: Author(s), Title, Edition, Publisher, Year of Publication etc. (Text books, Reference Books, Journals, Reports, Websites etc. in the IEEE format)	
1.	Experiment hand-outs.